PREPARATION OF TWO NEW BIS(2-PYRIDYL)SILANE LIGANDS AND THE COPOLYMERIZATION OF MESI(2-PYRIDYL)₂(CH=CH₂) WITH STYRENE AND DIVINYLBENZENE.

Michael E. Wright

Department of Chemistry & Biochemistry, Utah State University, Logan, UT 84322-0300

Summary: Reaction of two equiv of 2-(lithio)pyridine with one equiv of RMeSiCl $_2$ [R=Me(1a) & vinyl(1b)] produces the corresponding bis(2-pyridyl)silane in good yield. Compound 1a forms an air stable complex with PdCl $_2$ in which the bis(2-pyridyl)silane ligand is held in a rigid conformation. The suspension copolymerization of 1b with styrene and divinylbenzene is reported.

Chelating ligands are ubiquitous in the literature and are utilized in countless applications. The 2,2'-bipyridine ligands have commanded a considerable share of the attention over several decades. In more recent years, the interest in polymer-supported ligands has prompted the synthesis of 2,2'-bipyridine derivatives which contain a polymerizable vinyl group. In this paper we wish to describe a synthesis of two new bis(2-pyridyl)silane ligands which are easily prepared in a one-pot, two step procedure. In addition, the new ligands are shown to readily coordinate to palladium and the vinyl derivative can be incorporated with modest success into a polystyrene copolymer.

Reaction of 2-(lithio)pyridine² with either Me_2SiCl_2 or $Me(CH_2=CH)SiCl_2$ produces the corresponding Dis(2-pyridyl)Silanes, Dis(75%) and Dis(65%), respectively, in good yield after purification by distillation at reduced pressure.³

Treatment of $(CH_3CN)_2PdCl_2$ with one equiv of 1a in methylene chloride for 30 min produces the one:one complex 2 in quantitative yield (eq 2).⁴

$$Ia + (MeCN)_2 PdCI_2 \xrightarrow{CH_2CI_2} Me \xrightarrow{Si} Pd^{inCI}$$
(2)

Interestingly, the palladium complex displays two signals in the NMR spectra for the silyl methyl groups which we believe can best be rationalized in terms of a rigid chelate ring. In order for the ligand to accommodate the palladium the pyridine rings rotate about the pyridine-Si bond, which then destroys the C_2 axis and one of the mirror planes of symmetry. In the complex both pyridine rings are magnetically equivalent in $^1{\rm H}$ and $^1{\rm SC}$ NMR spectra which indicates a plane of symmetry remains which is defined by the Pd, Si, and the silyl-methyl groups. Heating a $^1{\rm H}$ NMR sample of 2 to 65° C showed no line broadening of the methyl resonances, indicating the energy barrier between the two possible isomers is reasonably large.

Suspension copolymerization⁵ of 1b(20 mol-3) with styrene and divinylbenzene gave polymer beads(20-50 μ m) containing approximately 1 mol-3 of the silane monomer as determined by nitrogen analysis. The lack of incorporation of the silyl monomer in the polymerization process indicates that the reactivity ratio is quite small in comparison to styrene and/or the silane monomer is destroyed under the polymerization conditions.

References

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- Yale, H. L. <u>Pyridine and Its Derivatives</u>, Part 2, Interscience, New York, 1961, Chapter 5.
- 3) la (bp 100°/.3 mm) ¹H NMR(CDCl₃) 8.80 (ddd, J= 1.5 Hz, 4.6 Hz, 2H), 7.56 (m, 4H), 7.20 (ddd, J= 5.9, 4.6, 3.2 Hz, 2H), 0.67 (s, 6H); ¹³C NMR(CDCl₃) 165.5, 150.1 133.9, 129.9, 122.9 (pyridine ring), -3.5 (SiCH₃). Anal Calcd. for C12H14N2Si: C, 67.25; H, 6.58. Found: C, 67.55; H, 6.72. lb (bp 115°/.2 mm) ¹H NMR(CDCl₃) 8.81 (m, 2H), 7.58 (m, 4H), 7.2 (m, 2H), 6.59 (dd, J= 14.6, 20.3 Hz, 1H), 6.26 (dd, J= 3.5, 14.6 Hz, 1H), 5.95 (dd, J= 3.5, 20.3 Hz, 1H), 0.77 (s, 3H); ¹³C NMR(CDCl₃) 164.0 (py C-2), 150.1, 150.2 (CH's), 135.6 (=CH₂), 133.9, 130.5, 129.9 (CH's), -5.5 (SiCH₃).
- 4) The complex was precipitated with ether, filtered, washed with cold chloroform, and dried at 40° under reduced pressure. 2: ¹H NMR (CDCl3) 9.23 (ddd, J= 5.7, 1.5, 0.9 Hz, 2H), 7.79 (apparent dt, J= 8.0, 1.5 Hz, 2H), 7.63 (ddd, J= 7.4, 1.7, 0.8 Hz, 2H), 7.37 (ddd, J= 7.6, 5.9, 1.7 Hz, 2H), 1.70 (s, 3H), 0.96 Ts, 3H); ¹³C NMR (CDCl3) 161.2 (py C-2), 155.7, 137.0, 130.5, 125.4 (pyridine CH's), -0.4, -4.5 (SiCH3).
- 5) The cross-linked copolymer was prepared using the procedure of: Parrinello, G.; Deschenaux, R.; Stille, J. K. J. Org. Chem. 1986, 51, 4189.

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